

PAPER • OPEN ACCESS

Application of Ozone Plasma Technology for Treating Peat Water into Drinking Water

To cite this article: F Ali *et al* 2019 *IOP Conf. Ser.: Earth Environ. Sci.* **385** 012056

View the [article online](#) for updates and enhancements.

Application of Ozone Plasma Technology for Treating Peat Water into Drinking Water

F Ali^{1,2,3}, D L Lestari², and M D Putri²

¹Department of Civil Engineering, Faculty of Engineering, Universitas Indonesia, Depok, 16424, Indonesia

²Indonesia Water Institute, Tanjung Barat Indah, South Jakarta, 12530, Indonesia

³Ministry of Public Works and Housing, Jl. Pattimura No. 20, Kebayoran Baru, South Jakarta, 12110, Indonesia

Email: firdausali@gmail.com

Abstract. The need of clean water will continue to increase along with the increase of population. However, not all regions in Indonesia have a potential for raw water sources that meet standards to be treated into clean/drinking water, so an alternative water treatment needs to be developed. Peat water is one of water sources that is often found in lowland swamps or in tidal conditions which is much located in Sumatera and Kalimantan. Peat water requires an appropriate technology to be able to treat water into clean water. Peat water treatment using ozone plasma technology is one of technological breakthroughs that can treat peat water into clean water effectively and economically. Peat water used for this study is taken from Sintang, West Kalimantan. Tested Parameters of water quality will be compared with the Regulation of Minister of Health No. 416 Year of 1990. In this study, four parameters in raw water are carried out to be tested after treating with variations in length of contact time of 10, 12, and 15 minutes. Water quality parameters to be tested are turbidity, color, coliform, and permanganate with final quality of treating for each parameter are 5.8 NTU, 1.1 Pt-Co, below 2 MPN/100 mL, and 11.8 mg/L. There is significant decrease in a value of four parameters, which three parameters meeting the maximum limit. Permanganate as parameter that exceed the maximum limit, have high probability of meeting this limit if contact time of treating peat water with ozone plasma is extended.

1. Introduction

Peat water in Indonesia is one of relatively abundant water resources. Results of the study of Geological Resource Center of Ministry of Energy and Mineral Resources report that until 2006 peat land resources in Indonesia covered a total of 26 million ha spread across Kalimantan (50%) and Sumatera (40%), while the rest were spread across Papua and other islands. Peat water has great potential to be treated into clean water, even with the advanced technology can be treated into drinking water. Peat water contains dissolved organic compounds that cause water to become brown and acidic, so it needs special treating before it can be consumed. Advanced technology is needed so that the purpose to utilize the peat water is optimal. Clean water treatment using disinfectants, filtering with membranes, ultra violet, and activated carbon does not guarantee that the water produced has a quality that is in accordance with the quality standards. This is evidenced by the results of some researches that microorganisms still exist in drinking water, especially in water produced in refill drinking water depots [1]. Currently, water treatment technology is actively being developed by oxidizing and decomposing organic pollutants in



water quickly, effectively and efficiently by using ozone (O_3). Ozone is known as one of the strongest oxidizers that can be produced in two ways: Electrical Discharge, Plasma Corona Discharge which is known as Ozone Plasma. Electrical Discharge is the formation of O_3 oxidizers through the O_2 ionization process using alternating electric current high voltage in the air [2]. Whereas, Plasma Corona Discharge is the formation of O_3 through the O_2 ionization process by using relatively high voltage electricity using low currents that are emitted transversely between plasma cells so that the production costs are relatively lower or cheaper to be applied in water treatment.

Ozone plasma in water can produce various kinds of active ion species that have the potential to decompose the content of organic compounds in water. In addition, plasma can also produce UV light and shock waves that have the potential to significantly decompose organic compounds in water [1]. Drinking water treatment with an ozone plasma system can replace the use of chlorine and activated carbon which has the function of killing microorganisms, and also at the same time can decompose the content of organic compounds. One development and application of ozone plasma technology for water treatment is to use ozone (O_3). In purifying water, ozone and hydroxide radicals formed in the ozone plasma system function as cleaning agents, strong oxidizers, disinfection sterilizers, deodorizers, self-purification, and even color decomposers. By looking at the ozone plasma working system and the characteristics of ozone, this study conducted an in-depth analysis to find out the benefits of using ozone plasma to treat peat water into clean/drinking water, especially in peat areas. This regarding that the use of ozone plasma technology has not implemented in the treatment of peat water, whereas in Indonesia there are many areas experiences clean water crisis, especially in areas that have raw water in the form of peat water.

2. Methodology

2.1. Research Method

This study uses independent variables and dependent variables. The independent variables are peat water sample originating from Sintang, West Kalimantan, and the length of time of contact (t_c) between water in ozone plasma unit. The dependent variables are the value of coliform, turbidity, color, and permanganate. The equipment used in this study is water purification unit with ozone plasma technology. This technology has been widely marketed and is commonly applied for treating clean water. With a series of experiments in the various of t_c , it will be known the effectiveness of the technology in treating the raw water (peat water) into drinking water.

2.2. Contact Mechanism and Ozone Plasma Reaction

Ozone plasma contact mechanism in treating raw water (peat water) is based on the chemical-physical operation process which leads to the formation of OH^- when the plasma unit is in contact with water (H_2O). Plasma units with a relatively high supply of electricity in water will produce ion particle charged with ultrasound energy that are resistant to UV and even shock waves, thus encouraging the formation of OH^- ions in solution [3]. In addition, the chemistry of plasma driven in the gas phase produces reactive oxygen and hydrogen peroxide which are not only radical, but also ozone, which will be needed to break down covalent gas so that it can produce plasma electrons. During the radical gas phase, there will be a diffusion process and changes in Henry's constant. Plasma also interacts with water which directly produces reactive ion species at the interface layer.

Interaction between ozone plasma and liquid will be able to provide an addition of oxidizers which will affect the oxidation level which is relatively much better than conventional methods, including: (1) consumables are not needed, because ozone plasma can be produced from ordinary air or the liquid itself, (2) the level of decomposition of organic compounds is relatively more effective, (3) because consumables are not needed, the costs and infrastructure normally used for consumables can be removed or substituted for the costs of overcoming other obstacles that may be encountered in water treatment, (4) ozone plasma application is inherently modular and can also be used as a final stage in water

treatment system that is no different from conventional UV stages for disinfection systems. The use of ozone plasma also allows for the elimination of micro contaminants.

In this ozone plasma system, peat water will be exposed directly with the concentration of O_3 produced through a plasma non-contact tube. By using this system, peat water is treated effectively and efficiently to produce clean water that is suitable for consumption. The mechanism for purifying water with an ozone plasma system begins by injecting oxygen into an area that has a high voltage. O_3 will be produced when passing through the corona plasma unit. When O_3 in the water content passes through a charged tube, radical hydroxide, an oxidizing compound, such as hydrogen peroxide is also produced in the presence of internal high stresses. With this mechanism there will be more production of effective hydroxide radicals. The oxidation ability possessed by O_3 and hydroxide radicals (OH^\cdot) will be able to destroy germs and decompose materials that cannot be degraded by conventional means. The ozone and hydroxide radicals that have been used will then break down into H_2O and O_2 forms which will then be released as treated water which has high dissolved oxygen and clarity that meet clean water standard.

In this study, an experimental treatment of peat water is conducted by varying the t_c , which is 10 minutes, 12 minutes, and 15 minutes. Time variations are made to determine the estimated effective time required by ozone plasma technology to reduce the parameters of coliform, turbidity, color, and permanganate which are the dominant parameters in peat water in general.

3. Results

In this study, tests are carried out on four water quality parameters, which are coliform, turbidity, color, and permanganate. Coliforms in water need to be checked because even though they are not dangerous to humans, coliforms are a good indicator of the presence of pathogenic microorganisms in water. Pathogenic microorganism is a variety of bacteria, parasites, and viruses that have the potential to endanger human health [4]. The presence of coliforms in water samples will be greatly considered because it can indicate the effectiveness of the disinfection process. Chlorine as a disinfectant in the treating of peat water cannot be used because in the chlorine disinfecting process, *trihalomethane* (THM) dangerous compounds will be formed which can be carcinogenic which are triggered by the higher solubility of metals in water because the pH of water is lower [5]. For peat swampy water, it is normal that pH and DO are generally low [6].

The treated water from ozone plasma treatment will be compared with the quality standard stated in the Regulation of The Minister of Health No. 416 Year of 1990 about Water Quality Requirements and Supervision. In testing the parameter of coliform, the raw water has a value of 13 MPN/100 mL, where the allowable limit is 10 MPN/100 mL. After 10 minutes of treating with ozone plasma technology, the coliform level is below 2 MPN/100 mL. By looking at the effectiveness of the ozone plasma technology on water purification, it can be seen that the disinfection process in the ozone plasma works very well.

Turbidity in water can be defined as a decrease in the transparency of a solution because of the presence of suspended or dissolved materials that can cause light to be dispersed, reflected and not transmitted straight. The higher the level of turbidity of water, the more light will be dispersed [7]. Turbidity in water is also caused by colloidal particles, and suspended particles, such as clay, silt soils, fine grains which are divided into organic and inorganic materials, plankton and microscopic organisms. Iron bacteria can also be a source of turbidity. The turbidity also acts as an important indicator for organic pollution, the run-off of suspended material and heavy rain fall in the area [8]. Turbidity removal can be done by giving a coagulant, which is then processed through coagulation and flocculation. In this study, the process is not carried out by given the magnitude of the ability of hydroxide radicals to decompose non-degraded material. This can be seen from the results of experiments that showed a decrease in turbidity level from the previous 50.15 NTU in raw water to 34.6 NTU in water treated with t_c 10 minutes, 12.6 NTU with t_c 12 minutes, and 5.84 NTU with t_c 15 minutes. When compared with the quality standard (25 NTU), it can be seen that purification of water with ozone plasma technology can meet quality standards while being able to cut the use of chemicals as a coagulant which commonly used in the water treatment process.

Natural colors in water occur because of the presence of complex organic molecules that are contained from humus substances such as leaves. The color in the water can also increase concentrated if there is the presence of suspended substances. Color on water usually affects aesthetics compared to health problems [4]. The water in peat swamp is almost black in color. The main reason is the presence of an organic material from peat decomposition [9]. Peat water has a very high color intensity, namely brownish red color, which is caused by a high content of organic matter dissolved mainly in the form of humus acid and its derivatives. Humus acid, derived from the decomposition of organic matter, such as leaves, trees, or wood with various levels of decomposition. In some cases, the color will be higher due to the presence of iron metal which is bound by organic acids dissolved in the water. Technique of color removal can be done in various ways, including: Coagulation-flocculation-sedimentation-filtration, oxidation-reduction, activated carbon, or adsorption-absorption. The technique of color removal basically requires additional media or chemicals that will have a significant effect on the cost of water treatment. In this research, no additional media or chemicals are used at all, only by maximizing the processes that occur in ozone plasma system, where O_3 and OH^\cdot dissolved in water will function to remove color. The color content of the raw water of peat water shows value of 7.8 Pt-Co with maximum limit according to the Regulation of The Minister of Health No. 416 Year of 1990 is 50 Pt-Co. With t_c 10minute, treating the water with ozone plasma unit achieves 3.5 Pt-Co and if t_c is increased to 15 minutes, the color level of treated water reaches 1.1 Pt-Co. This shows that treating peat water using ozone plasma has been proven to reduce the color level in the water.

Permanganate number is used to measure the high content of organic matter in water. Permanganate value is the amount of milligram of potassium permanganate needed to oxidize organic matter in 1000 mL of water in boiling conditions [10]. Excessive organic matter in drinking water is not permitted because in addition to causing unwanted colors, smells and tastes, it may also be toxic both directly and after compounding with other existing substances [11]. Permanganate number is closely related to the value of COD (Chemical Oxygen Demand), BOD (Biochemical Oxygen Demand), and DO (Dissolved Oxygen). The relationship between BOD and COD values with permanganate numbers are the three parameters which together indicate the number of organic (and non-organic) compounds dissolved in the waters. Thus, the higher the permanganate number, the higher the COD value and the BOD value.

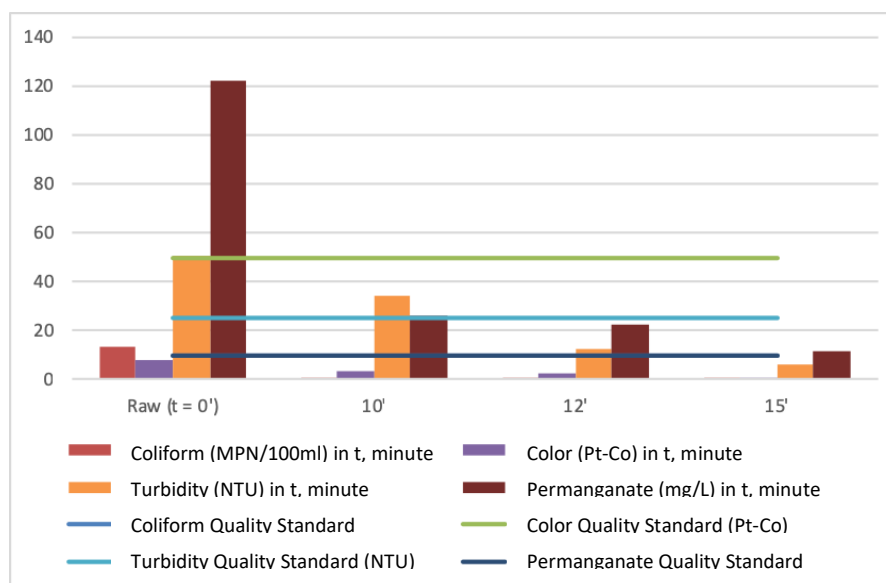


Figure 1. Results of Peat Water Treatment with Ozone Plasma Technology

The more complex water treatment system is needed if the BOD and COD content is high. The permanganate number in the peat water is 122.3 mg/L with allowable limit is 10 mg/L. In this study, it is found that after 10 minutes of treating, the permanganate number can be reduced to 26.3 mg/L. By using t_c 12 minutes, the permanganate number can be reduced to 22.8 mg/L. By extending t_c 3 minutes which is for total of 15 minutes of treating, the final value of permanganate number drops by almost half, reaching 11.8 mg/L. By looking at the results of this experiment, the treating experiments with t_c 15 minutes are able to significantly reduce the level of permanganate in peat water close to the quality standard. It can be said that the permanganate method in acidic condition has a better recovery [12]. This is probably due to the fact that most of the organic matters were oxidized more completely in acidic than in alkaline conditions [13]. To obtain better permanganate numbers, treating by the ozone plasma technology is needed with a t_c of more than 15 minutes or an estimated treating time of no more than 20 minutes. The results of this experimental treatment using ozone plasma technology are showed in Figure 1.

4. Discussion

Peat water which is widely found on the islands of Borneo and Sumatra is a type of water that is commonly found in locations that often occur in tidal conditions, swampy areas, and lowlands. In general, peat water has acidic and brown properties and based on the Government Regulation No. 82 Year of 2001 concerning management of water quality and water pollution control, defines peat water as unsuitable for consumption so it must be treated in advance so that it is suitable for consumption [14], while swampy areas generally only have surface water from peat water. One appropriate solution is to use water treatment plant that can treat the peat water into consumable water. But conventional technology is not appropriate to be used to treat peat water. This is because to treat peat water into consumable water requires high technology and expensive costs so that its affordability by the community will be very low. Water treatment technology using ozone plasma is proven to be able to treat peat water into consumable water.

After treating the peat water by ozone plasma technology with t_c 15 minutes, three parameters tested (coliform, color, and turbidity) are proven to be able to be treated and to meet the clean water quality standards based on the Regulation of The Minister of Health No. 416 Year of 1990. However, there is one parameter that has not met the maximum limit of clean water, which is permanganate number. Even though at t_c 15 minutes the permanganate number is still above the maximum limit, there is a significant decrease and it is highly probable that permanganate number can be below the maximum limit if it is treated with a longer contact time.

5. Conclusion

Based on the results of treating peat water using ozone plasma technology, it is proven that the technology able to treat peat water very effectively compared to the conventional refining technology. At the t_c 15 minutes treating peat water with ozone plasma, 3 of 4 test parameters (coliform, color, and turbidity) can be treated to be below the maximum limit. While another test parameter can be reduced significantly so the value is only slightly exceeding the maximum limit, and there is a high probability that the value can be treated to below the maximum limit by increasing the contact time of peat water in the ozone plasma unit. The use of ozone plasma to treat peat water will be able to reduce the use of chemicals to treat water into clean water so that it will have a significant impact on the cost of treating peat water. Thus, the use of this ozone plasma technology to treat peat water will be an effective cost to be developed and applied in many places on the islands of Kalimantan and Sumatra in Indonesia. This regarding that in the areas that have raw water in the form of peat water, the use of ozone plasma technology can be one of appropriate alternative to treat the raw water into clean/drinking water.

Acknowledgment

This research can be carried out because of the support of tool facilities, laboratories, and staff from Teknikon Surabaya. Thank you to Dwi Lintang Lestari and Marsya Dyasthi Putri from Indonesia Water Institute-Jakarta who have helped with the research and writing of this paper.

References

- [1] Hazmi, A., Desmiarti, R., and Waldi, E.P. (2011). Penghilangan Mikroorganisme dalam Air Minum dengan Pulsa Tegangan Tinggi (*Removal of Microorganisms in Drinking Water with High-Voltage Pulses*). Electrical Engineering, Faculty of Engineering, Andalas University.
- [2] Yuan, D., Wang, Z., He, Y., Xie, S., Lin, F., Zhu, Y., and Cen, K. (2018). Ozone Production with Dielectric Barrier Discharge from Air: The Influence of Pulse Polarity. *The Journal of the International Ozone Association*, 40 (6): 494-502.
- [3] Foster, J.E. (2017). Plasma-based water purification: Challenges and prospects for the future. *Physics of Plasmas*, 24 (5), 055501.
- [4] Environmental Protection Agency. (2001). *Parameters of Water Quality: Interpretation and Standards*. Ireland: Author.
- [5] Richardson, S., Plewa, M., Wagner, E., Schoeny, R., and Demarini, D. (2007). Occurrence, genotoxicity, and carcinogenicity of regulated and emerging disinfection by-products in drinking water: a review and roadmap for research. *Mutation Research/Reviews in Mutation Research*, 636 (1-3), 178-242.
- [6] Gasim, M.B., Ismail, B.S., Toriman, E., Mir, S.I., and Chek, T.C. (2007). A Physico-Chemical Assessment of the Bebar River, Pahang, Malaysia. *Global Journal of Environmental Research*, 1 (1): 07-11.
- [7] Ziegler, A.C. (2002). Issues Related to Use of Turbidity Measurements as a Surrogate for Suspended Sediment. *Turbidity and Other Sediment Surrogates Workshop, April 30 – May 2, 2002, Reno, NV*.
- [8] Yisa, J., and Jimoh, T. (2010). Analytical studied on water quality index of River Landzu. *American Journal of Applied Sciences*, 7 (4): 453-458.
- [9] Rosli, N., Gandaseca, S., Ismail, J., and Jailan, M.I. (2010). Comparative Study of Water Quality at Different Peat Swamp Forest of Batang Igan, Sibul Serawak. *American Journal of Environmental Sciences*, 6 (5): 416-421.
- [10] Badan Standarisasi Nasional. (2004). *SNI 06-6989.22-2004 Air dan air limbah – Bagian 22: Cara uji nilai permanganat secara titrimetri* (Indonesian National Standard 06-6989.22-2004 Water and wastewater – Part 22: How to test the permanganate value by titrimetry). Serpong: Author.
- [11] Soesanto, S.S. (1996). Senyawa Organik Dalam Air Minum (*Organic Compounds in Drinking Water*). *Media Penelitian dan Pengembangan Kesehatan* (Media of Health Research and Development), Vol. VI No. 01.
- [12] Goh, C.P., and Lim, P.E. (2006). Potassium permanganate as oxidants in the COD test for saline water samples. *ASEAN Journal on Science and Technology for Development*, 25 (2): 383-393.
- [13] Fujimori, K., Ma, W., Kawakami, T.M., and Shibutani, T. (2001). Chemiluminescence Method with Potassium Permanganate for the Determination of Organic Pollutants in Seawater. *Analytical Sciences*, Vol. 17.
- [14] Juhra, F., and Notodarmojo, S. (2016). Degradasi Zat Warna pada Air Gambut Menggunakan Metode Fotokatalik ZnO (*Degradation of Dyes on Peat Water Using ZnO Photocatalytic Method*). *Jurnal Teknik Lingkungan* (Journal of Environmental Engineering), 2 (2): 1-10.
- [15] Environmental Protection Agency. (2009). National Primary Drinking Water Regulations. *EPA 816-F-09-004*.